Sl.No: M21413 Course Code: BP104T

# VINAYAKA MISSION'S RESEARCH FOUNDATION (DEEMED TO BE UNIVERSITY), SALEM

## B.PHARM. DEGREE EXAMINATION – JULY 2019 First Semester

#### PHARMACEUTICAL INORGANIC CHEMISTRY

Time: Three hours Maximum: 75 marks

I. Write essays on any **TWO** questions:

 $(2 \times 10 = 20)$ 

- 1. Write the principle, procedure and reaction involved in limit test for heavy metals and sulphates.
- 2. Describe the method of preparation, assay and uses of ammonium chloride.
- 3. What is physiological acid base balance? Explain the preparation and assay of calcium gluconate.

### II. Write short answers on any **SEVEN** questions:

 $(7 \times 5 = 35)$ 

- 4. What are expectorants? Write the preparation and uses of potassium iodide.
- 5. What is radioactivity? Explain any one method to measure the radioactivity.
- 6. Explain the mechanism of action of antimicrobial agents. Write the preparation of boric acid.
- 7. What is antacid combination therapy? Write the principle involved in the assay of sodium bicarbonate.
- 8. Define and classify dental products. Explain the role of fluorides in the treatment of dental caries.
- 9. Write notes on chlorinated lime.
- 10. Explain the physiological role of potassium and chloride ions.
- 11. Write the applications of radioactive substances.
- 12. Write the preparation, properties and uses of magnesium sulphate.

#### III. Write short notes on:

 $(10 \times 2 = 20)$ 

- 13. Define cathartics with an example.
- 14. What are astringents? Write the use of zinc sulphate.
- 15. Define emetics. Write the molecular formula of sodium potassium tartrate.
- 16. What are radioisotopes? Give examples.
- 17. Write a note on kaolin.
- 18. Write the procedure involved in the limit test for chloride.
- 19. Write the preparation method of aluminium hydroxide gel.
- 20. Write the storage condition and uses of copper sulphate.
- 21. Define dentifrices with examples.
- 22. Write about the assay of hydrogen peroxide.